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AROMADENDRENE AND VIRIDIFLOROL

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FORMULA (I) advanced ¹ as a possibility for aromadendrene is more likely to be correct than the alternative with the <u>cyclopropane</u> ring terminating at the 6-position since dehydrogenation gives only **S**-guaiazulene. Ozonolysis of aromadendrene from <u>Metrosideros scandens</u> ² gave a-apoaromadendrone, isomerised by alkali to the well known apoaromadendrone, and it was assumed that this hydrocarbon and the aromadendrene from <u>Eucalyptus</u> <u>nova-anglica</u> and other sources were identical and therefore had a <u>trans</u>-9,10-ring junction. Failure to obtain a-apoaromadendrone from the latter hydrocarbon in the Manchester and several other laboratories indicates

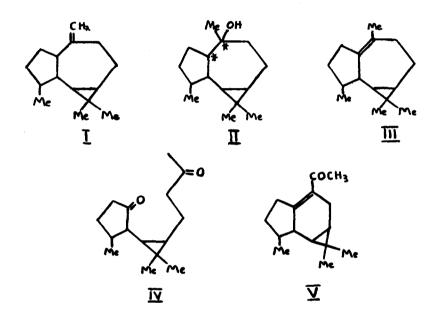
¹ A.J. Birch and F.N. Lahey, <u>Aust. J. Chem.</u> <u>6</u>, 379 (1953).

² A.J.Birch, <u>J. Chem. Soc.</u> 715 (1953).

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that the <u>M</u>. <u>scandens</u> hydrocarbon (a-aromadendrene) is a stereoisomer with a <u>trans</u>-9,10-ring junction and that the authentic aromadendrene has the <u>cis</u>-ring junction.

Globulol has been shown 3 to be a stereoisomer of (II) related stereochemically to aromadendrene; we have found that viridiflorol 4 (identical with himbaccol 5)



- ³ A. Blumann, A.R.H. Cole, K.J.L. Thiaberg and D.E. White, <u>Chem. and Ind.</u> 1426, (1954).
 ⁴ T.G.H. Jones and W.L. Haenke, <u>J. Roy. Soc.</u> <u>Queensland</u> 49, 95 (1938).
- ⁵ A.J. Birch and K.M.C. Mostyn, <u>Aust. J. Chem.</u> <u>8</u>, 550 (1955).

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yields some a-aromadendrene on pyrolysis of its phenylurethane. The product (C=CH₂ bands at 3085; 1638; 890 cm⁻¹) on ozonolysis gave a-apoaromadendrone, m.p. 71-72°; $[\alpha]_{n}^{23}$ -l¹⁶, isomerised by base to apoaromadendrone, m.p. and mixed m.p. $83-84^{\circ}$; $[a]_{n}^{23}+3^{\circ}$. Dehydration of viridiflorol by alcoholic sulphuric acid produced some ledene identified as the glycol, ⁶ m.p. and mixed m.p. 150-151°; $[\alpha]_D^{20}$ -15.5°. Lead tetraacetate fission of the glycol gave a diketone (IV) (carbonyl bands at 1740 and 1715 cm⁻¹, bis-2,4-dinitrophenylhydrazone, m.p. 115-118°) also obtained by chromic acid oxidation of viridiflorol. The diketone was cyclised by alkali to an $\alpha\beta$ -unsaturated ketone probably (V) (acetyl $\boldsymbol{\nu}_{max}$ 1356 cm⁻¹) contaminated with its β **8**-isomer $(\lambda_{\max}^{\text{alc.}} 24 \text{lm}\mu; \boldsymbol{\nu}_{\max} 1710 \text{ and } 1688 \text{ cm}^{-1})$. This supports the formula (III) advanced for ledene⁷ and the suggestion that ledol and palustrol ⁸ are probably stereoisomers of (II).⁷ Therefore in all probability ledol, palustrol,

- ⁶ N.P. Kyrialov, <u>Zhur. Obshcheĭ Khim. 21</u>, 2077 (1951); <u>Chem. Abs. 46</u>, 6633 (1952)
 ⁷ L. Dolejs, F. Sorm and M. Soucek, <u>Chem. and</u>
- Ind. 160 (1959).
- ⁸ N.P. Kyrialov, <u>Zhur. Obshcheľ Khim.</u> <u>24</u>, 1271 (1954); <u>Chem. Abs.</u> <u>49</u>, 13944 (1955).

Aromadendrene and viridiflorol

viridiflorol and globulol are the four possible diastereoisomers of (II) differing at the asterisked positions. Professor F. Šorm (Prague), to whom we are indebted for a specimen of ledeneglycol, has informed us that he and his co-workers have reached similar conclusions.

We have confirmed the observation of Kyrialov ⁹ that ledeneglycol on acid treatment followed by dehydrogenation yields 1,6-dimethyl-1-isopropylnapth-5-ol the picrate of which was identical with an authentic specimen prepared from guaiol.¹⁰ The weight of evidence in favour of structure (II) for viridiflorol is so great that the mechanism of this rearrangement must be more complex than Sorm originally proposed.¹¹

- ⁹ N.P. Kyrialov, <u>Sbornik Statei Obshcheľ Khim.</u>
 <u>2</u>, 1617 (1953); <u>Chem. Abs.</u> <u>49</u>, 5389 (1955).
- ¹⁰ Pl.A. Plattner and G. Magyar, <u>Helv.Chim. Acta</u>, <u>24</u>, 191; 1163 (1941).
- 11 L. Dolejs, M. Soucek, M. Horak and F. Sorm, <u>Chem. and Ind.</u> 494 (1958); <u>Chem. Listv.</u>, <u>52</u>, 2188 (1958).

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